

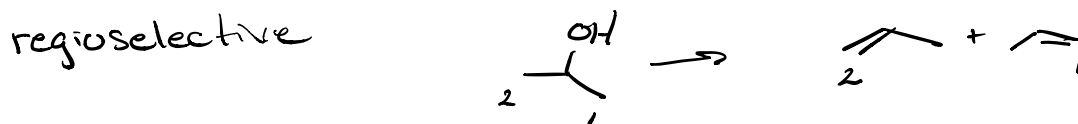
# Elimination Rxns

$E_1$  Elimination 1<sup>st</sup> order

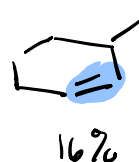
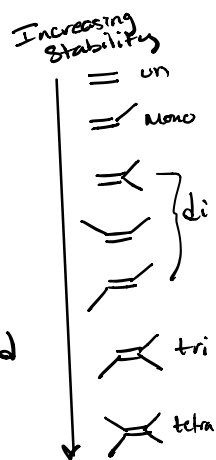
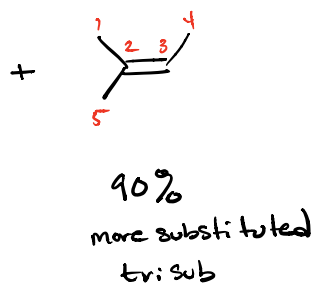
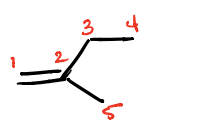
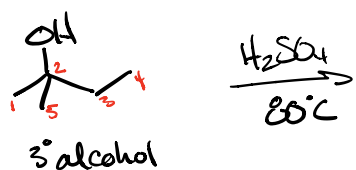
$E_2$  Elimination 2<sup>nd</sup> Order



dehydration Rxn  
both regio & stereoselective

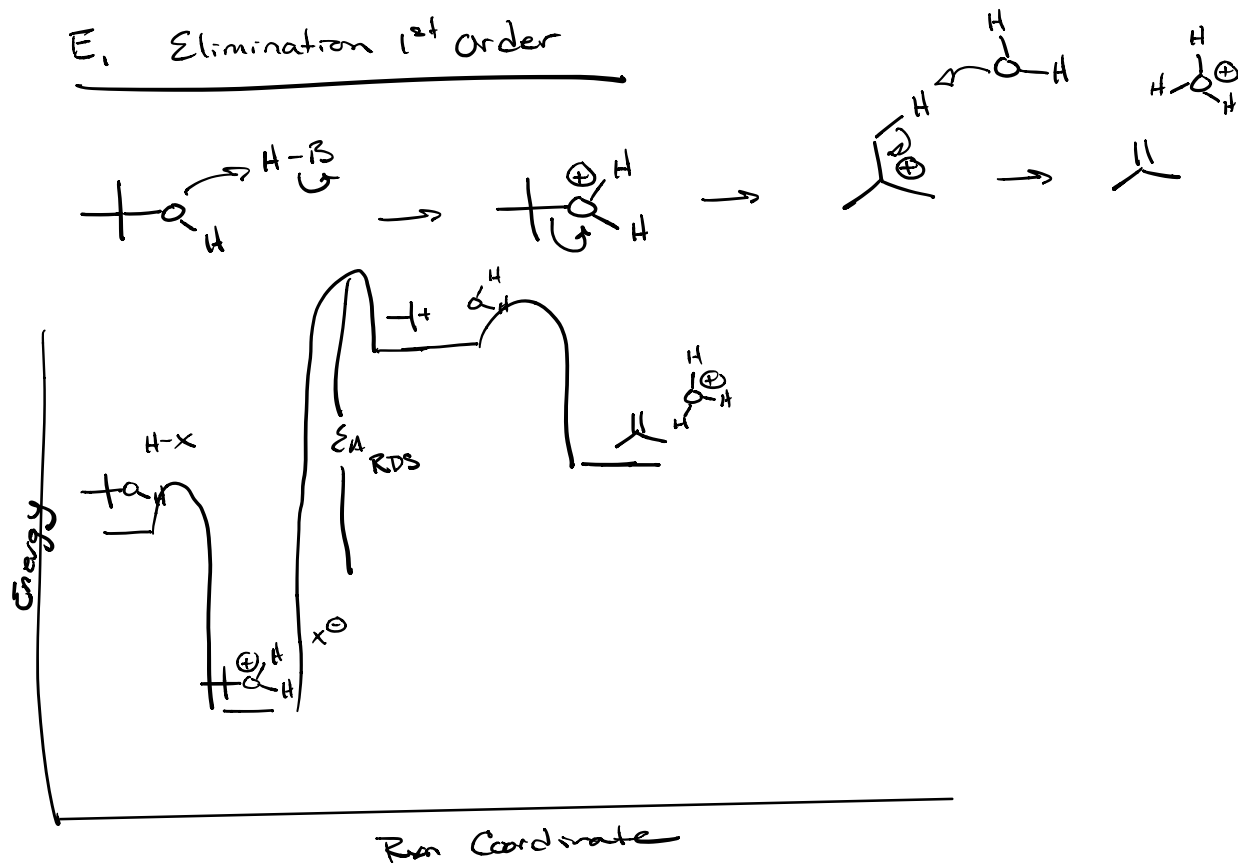


## Regioselective



# Zaitsev's Rule (1875 Alexander Zaitsev)

E<sub>1</sub> Elimination 1<sup>st</sup> Order

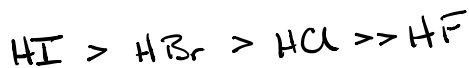


$$\text{Rate} = k [\text{R-LG}] \Rightarrow \text{Identical Sulf}$$

Same factors apply as for S<sub>N</sub>1

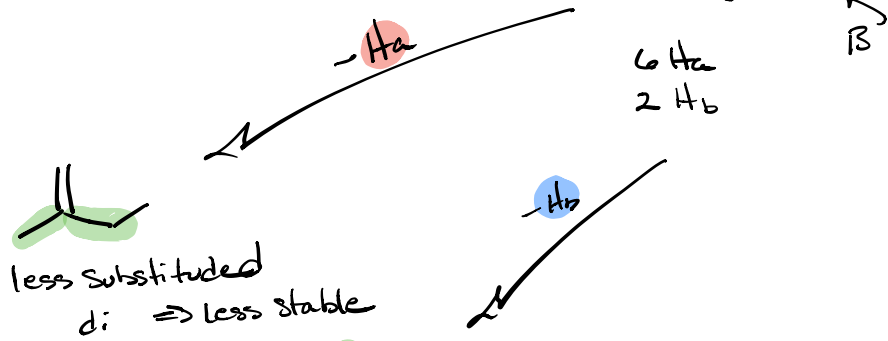
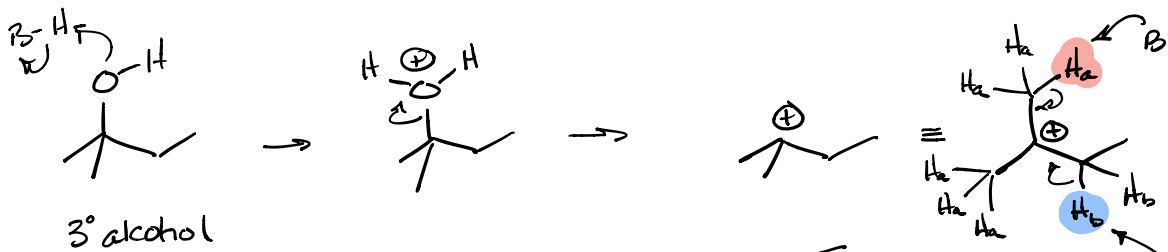
$$\text{Rate } 3^\circ > 2^\circ > 1^\circ$$

Substrate  
R-LG

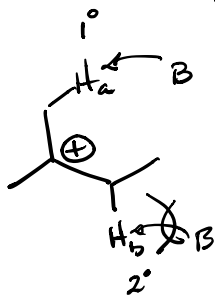


←  
Stronger acid

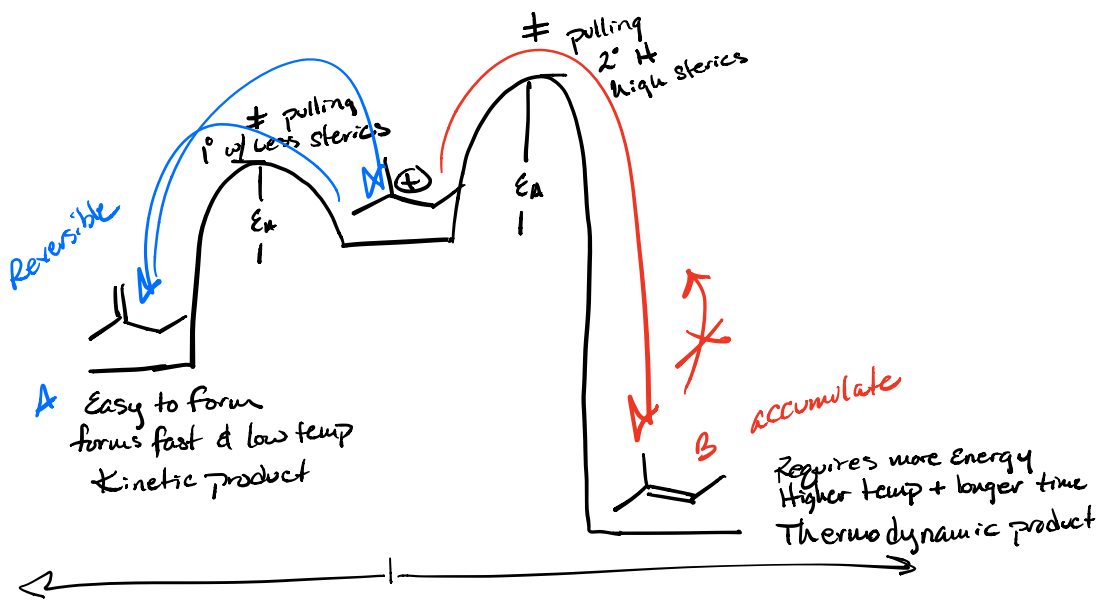
$$\text{Rate} = k [\text{R-OH}_2^+]$$

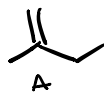
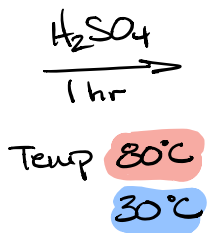
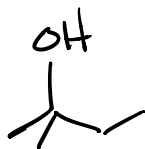


$1^\circ$  less sterics  
 to access the  
 $1^\circ$  proton  $E_a \downarrow$



$2^\circ$  more sterics  
 around  $2^\circ$  proton  $E_a \uparrow$



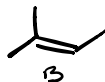


10%

**60%** *Reversible*

Kinetic

low temp  
short Run time

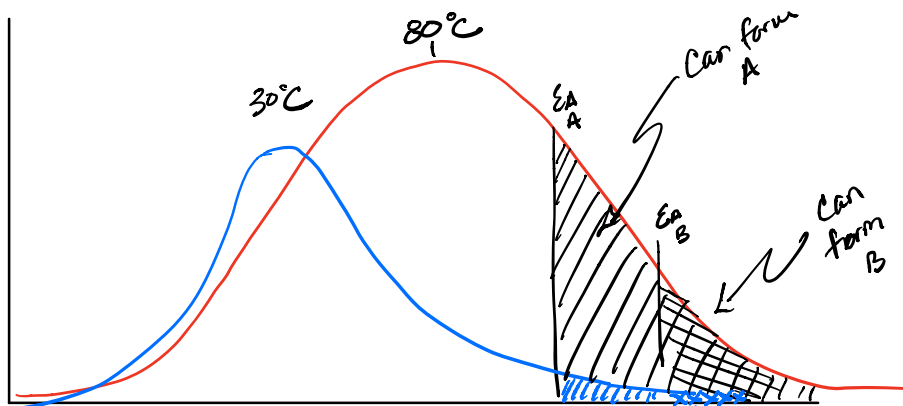


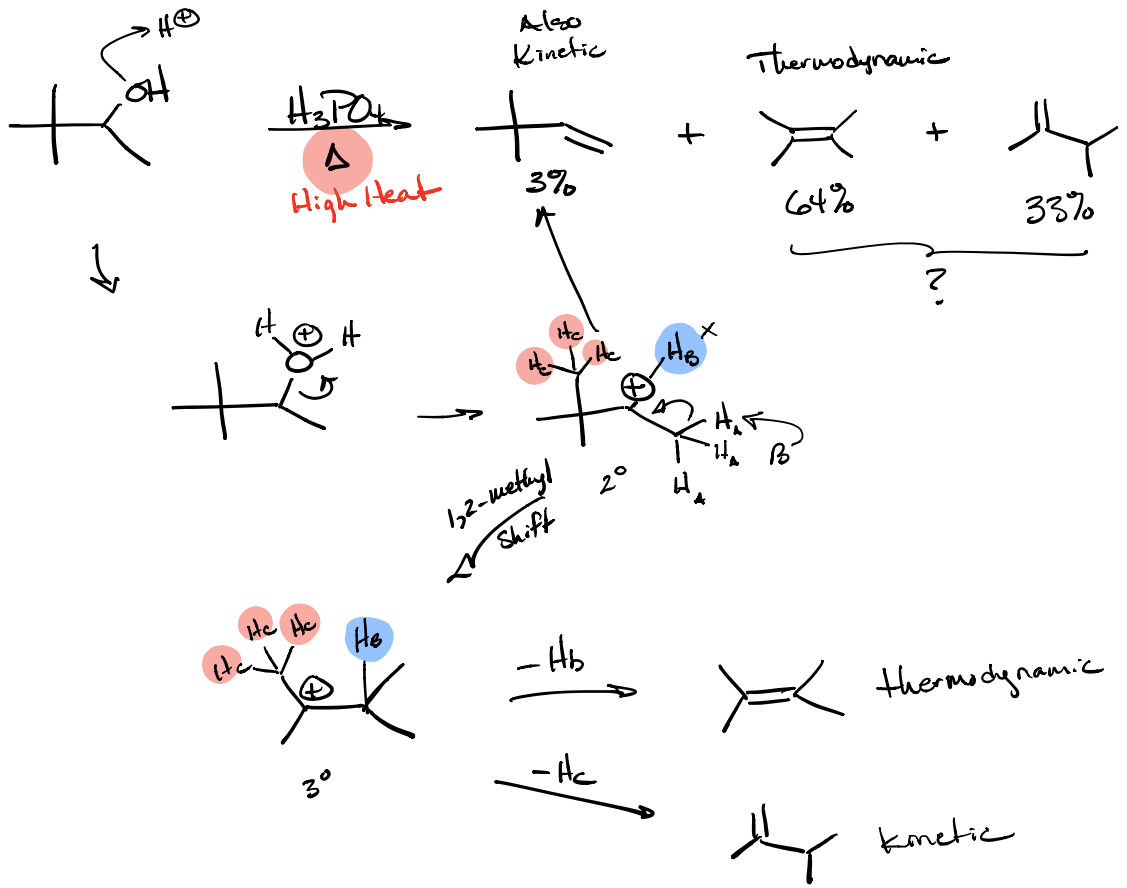
**90%** ← *accumulating*

40%

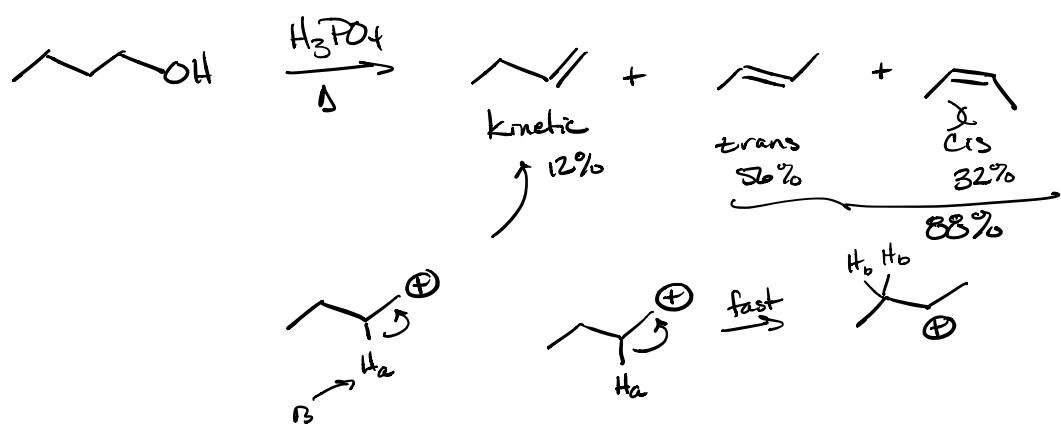
Thermodynamic

High temp  
long Run time

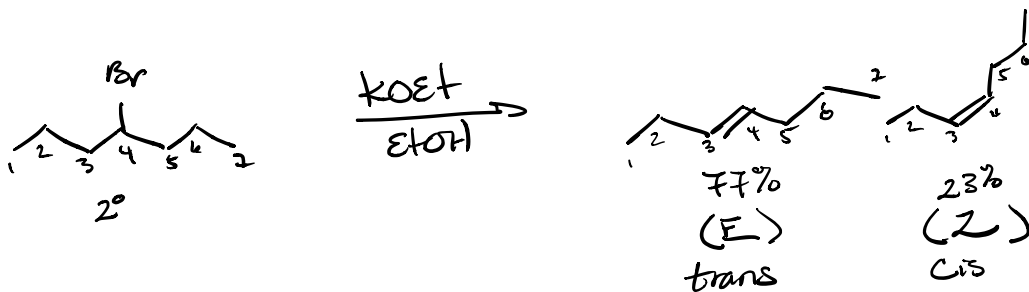
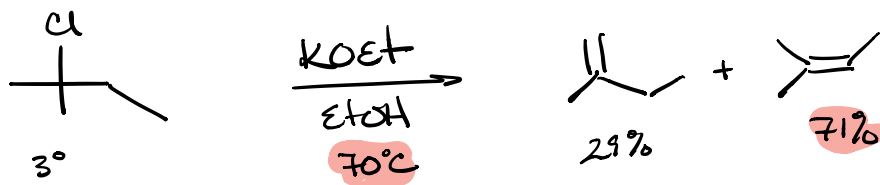
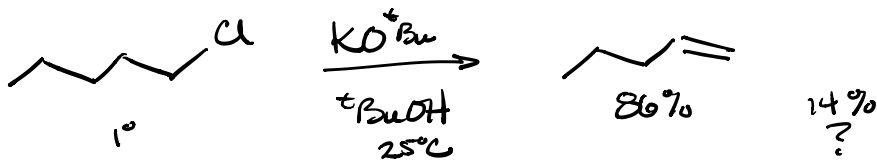
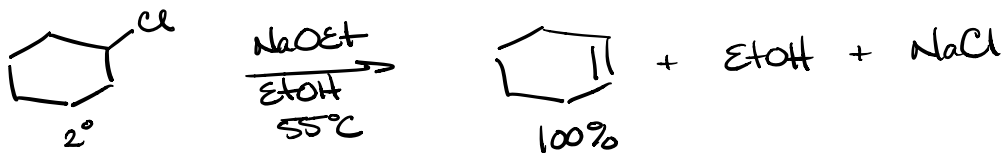




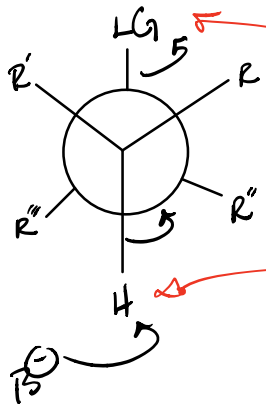
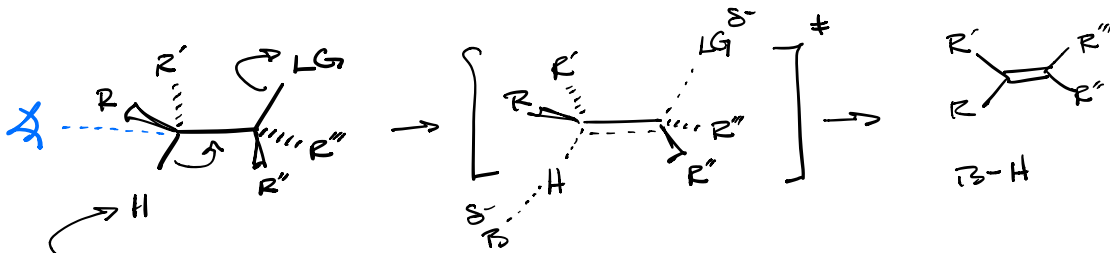
$\Delta$  = Reflux = High heat = Thermodynamic product



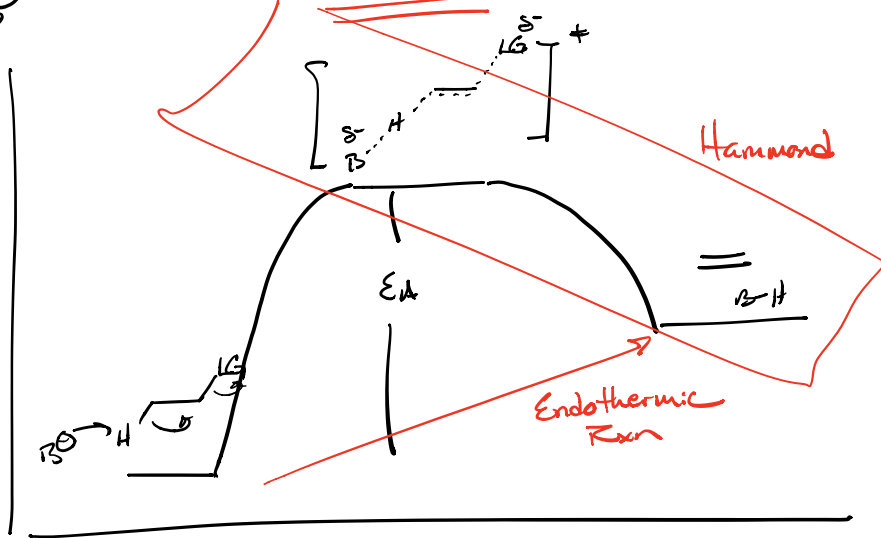
# Dehydrohalogenation



# E<sub>2</sub> Elimination 2<sup>nd</sup> Order



H & LG  
anti  
to one  
another



Concerted = Bond breaking &  
Bond making  
Simultaneous

$$E_2 \text{ Rate} = k [\text{Base}] [\text{R-LG}]$$

$$S_N2 \text{ Rate} = k [\text{nuc}] [\text{R-LG}]$$

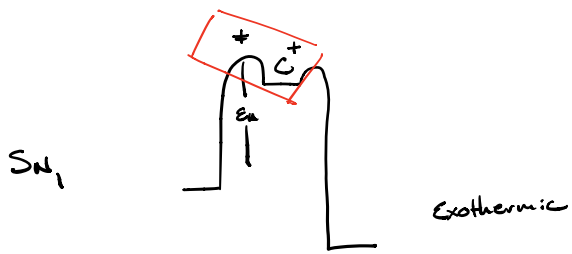
difference Base vs. Nucleophile

$E_2$  base

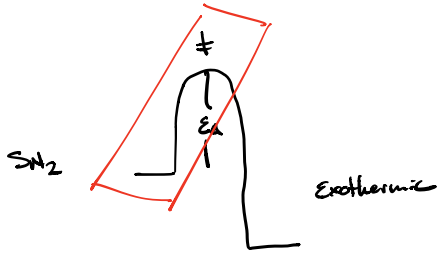
$S_N2$  nucleophile

Rate  $3^\circ > 2^\circ > 1^\circ \Rightarrow$  # product like  
R-LG structure of the product  
(stability of product)  
governs Rxn Rate

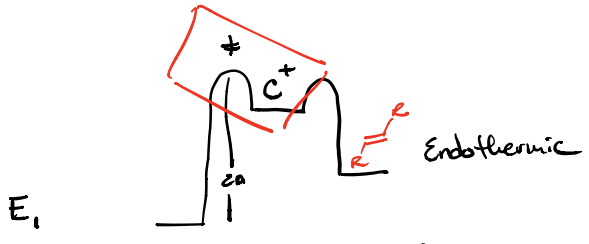




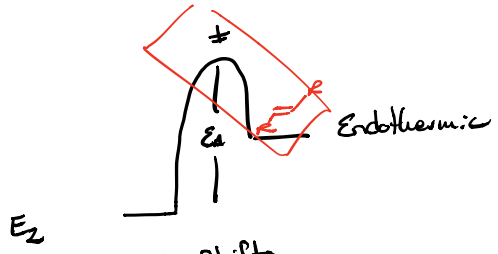
Alkyl & hydride shifts  
Intermediate  $C^+$   
Stepwise



no shifts  
no intermediate  
Concerted

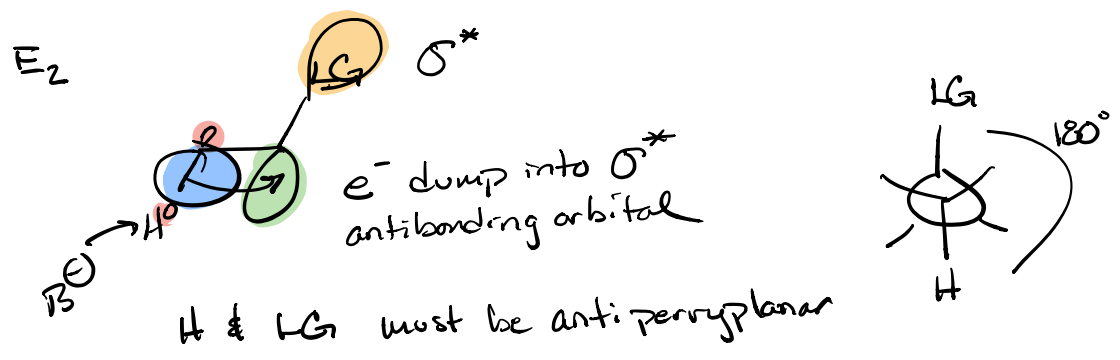
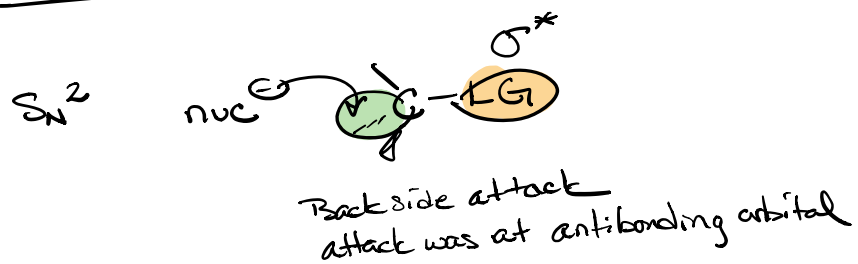


Alkyl & hydride shifts  
Intermediate  $C^+$   
Stepwise

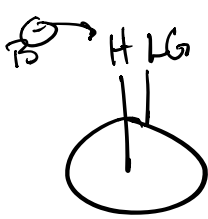


no shifts  
no intermediate  
Concerted

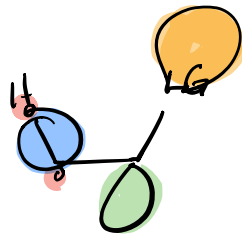
### Stereoelectronic Effects



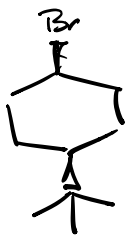
When Syn Coplanar



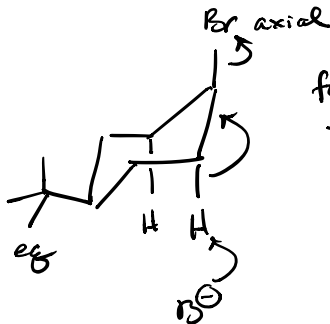
=



→ no Rxn  
or at least  
very very slow



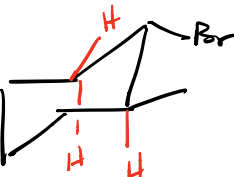
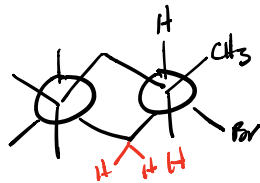
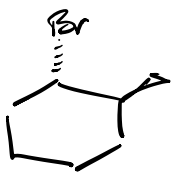
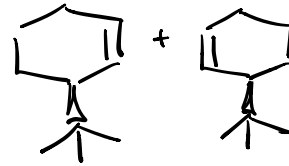
NaOEt  
EtOH  
E<sub>2</sub>



fast



H & Br locked  
anti due to  
t-Bu group

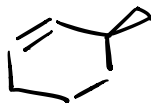


Major  
Conformation  
lowest in  
energy  
no anti  
hydrogens

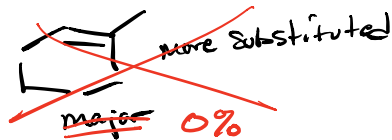
only Conformation  
that can undergo  
E<sub>2</sub> Rxn

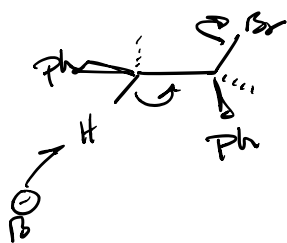


NaOEt  
EtOH



only product  
because only 1 Anti H can be pulled





H & LG  
must be  
anti:  
    

